National University of Technology, Islamabad
Quiz I (Linear Algebra \& ODE), Fall 2019
BET (Mechanical), Date: September 26, 2019

Q Balancing a chemical equation $w C_{3} H_{8}+x O_{2} \rightarrow y \mathrm{CO}_{2}+z \mathrm{H}_{2} \mathrm{O}$ means finding integers $w, x, y, z$ such that the numbers of atoms of carbon ( C ), hydrogen ( H ) and oxygen ( O ) are the same on both sides of this reaction, in which propane $C_{3} H_{8}$ and $O_{2}$ give carbon dioxide and water. Find the smallest positive integers $w, x, y$ and $z$ using Echelon or Reduced Echelon Form. (Hint: Compare the atoms of C,H and $O$ on both sides).
"Everything looks impossible to the people who never try anything."

